

REFERENCE SHEET

ELECTRONEGATIVITIES

H																
2.20																
Li	Be											B	C	N	O	F
0.98	1.57											2.04	2.55	3.04	3.44	3.98
Na	Mg											Al	Si	P	S	Cl
0.93	1.31											1.61	1.90	2.19	2.58	3.16
K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br
0.82	1.00	1.60	1.54	1.63	1.66	1.55	1.83	1.88	1.91	1.90	1.65	1.81	2.01	2.18	2.55	2.96
Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I
0.82	0.95	1.22	1.33	1.60	2.16	2.10	2.20	2.28	2.20	1.93	1.69	1.78	1.96	2.05	2.10	2.66
Cs	Ba	La	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At
0.79	0.89	1.10	1.30	1.50	1.70	1.90	2.20	2.20	2.20	2.40	1.90	1.80	1.80	1.90	2.00	2.20
Fr	Ra	Ac														
0.70	0.90	1.10														

Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	No	Er	Tm	Yb	Lu
1.12	1.13	1.14		1.17		1.20		1.22	1.23	1.24	1.25		
Th	Pa	U	Np	Pu									
1.30	1.50	1.70	1.30	1.30									

from *Solving Problems, A Chemistry Handbook*

CATIONS

aluminum Al^{+3}
 ammonium NH_4^{+1}
 antimony (III) Sb^{+3}
 antimony (IV) Sb^{+5}
 barium Ba^{+2}
 bismuth (III) Bi^{+3}
 bismuth (V) Bi^{+5}
 calcium Ca^{+2}
 cadmium Cd^{+2}
 chromium (II) Cr^{+2}
 chromium (III) Cr^{+3}
 cobalt (II) Co^{+2}
 cobalt (III) Co^{+3}
 copper (I) Cu^{+1}
 copper (II) Cu^{+2}
 hydrogen H^{+1}
 iron (II) Fe^{+2}
 iron (III) Fe^{+3}
 lead (II) Pb^{+2}
 lead (IV) Pb^{+4}
 lithium Li^{+1}
 magnesium Mg^{+2}
 manganese (II) Mn^{+2}
 manganese (IV) Mn^{+4}
 mercury (I) Hg_2^{+2}
 mercury (II) Hg^{+2}
 nickel (II) Ni^{+2}

nickel (III) Ni^{+3}
 potassium K^{+1}
 silver Ag^{+1}
 sodium Na^{+1}
 strontium Sr^{+2}
 tin (II) Sn^{+2}
 tin (IV) Sn^{+4}
 zinc Zn^{+2}

ANIONS

acetate $\text{C}_2\text{H}_3\text{O}_2^{-1}$
 arsenate AsO_4^{-3}
 bicarbonate HCO_3^{-1}
 binoxalate $\text{HC}_2\text{O}_4^{-1}$
 biphosphate HPO_4^{-2}
 bisulfate HSO_4^{-1}
 bisulfide HS^{-1}
 bisulfite HSO_3^{-1}
 borate BO_3^{-3}
 bromide Br^{-1}
 carbonate CO_3^{-2}
 chlorate ClO_3^{-1}
 chloride Cl^{-1}
 chlorite ClO_2^{-1}
 chromate CrO_4^{-2}
 citrate $\text{C}_6\text{H}_5\text{O}_7^{-3}$
 cyanide CN^{-1}

dichromate $\text{Cr}_2\text{O}_7^{-2}$
 dihydrogen
 phosphate $\text{H}_2\text{PO}_4^{-1}$
 ferrocyanide $\text{Fe}(\text{CN})_6^{-4}$
 ferricyanide $\text{Fe}(\text{CN})_6^{-3}$
 fluoride F^{-1}
 hydroxide OH^{-1}
 hypochlorite ClO^{-1}
 iodate IO_3^{-1}
 iodide I^{-1}
 molybdate MoO_4^{-2}
 nitride N^{-3}
 nitrate NO_3^{-1}
 nitrite NO_2^{-1}
 oxalate $\text{C}_2\text{O}_4^{-2}$
 oxide O^{-2}
 perchlorate ClO_4^{-1}
 permanganate MnO_4^{-1}
 peroxide O_2^{-2}
 phosphate PO_4^{-3}
 phosphide P^{-3}
 silicate SiO_3^{-2}
 sulfate SO_4^{-2}
 sulfide S^{-2}
 sulfite SO_3^{-2}
 tartrate $\text{C}_4\text{H}_4\text{O}_6^{-2}$
 thiocyanate SCN^{-1}

ACTIVITY SERIES

metals	nonmetals
Li	F
K	O
Ba	Cl
Sr	Br
Na	I
Ca	
Mg	
Al	
Mn	
Zn	
Cr	
Fe	
Cd	
Co	
Ni	
Sn	
Pb	
H	
Sb	
As	
Bi	
Cu	
Hg	
Ag	
Au	